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relationship between the nu.... It applies only to gases measured under the same temperature a....

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334 CHAPTER 11 FIGURE 11-1 At the same temperature and pressure, balloons of equal volume have equal numbers of molecules, regardless of which gas they contain. Hydrogen molecule 1 mol H<sub>2</sub> at STP = 22.4 L  
Oxygen molecule 1 mol O<sub>2</sub> at STP = 22.4 L  
Carbon dioxide molecule 1 mol CO<sub>2</sub> at STP = 22.4 L

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Answer Key Chapter 11 - Chemistry 2e | OpenStax 1. A solution can vary in composition, while a compound cannot vary in composition. Solutions are homogeneous at the molecular level, while other mixtures are heterogeneous.

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