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Sodium Fluoride Goes To School

Fluoridation involves adding sodium fluoride, hydrofluorosilicic acid, or sodium fluorosilicate to water. Hydrofluorosilicic acid is made when phosphoric rock is converted to soluble fertilizer. The powder version of this, called sodium fluoride, is one of the main ingredients in pesticides, insecticides, and even fungicides.

Fluoridation or Contamination? | Education

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The clean caustic leachate goes to an evaporator-crystallizer where the concentration of sodium hydroxide is raised to a point (around 225 g/L) to crystallize out sodium fluoride. As the leachate is further evaporated, the solubility of sodium fluoride is further reduced as sodium carbonate and alumina start to coprecipitate.

Sodium Fluoride - an overview | ScienceDirect Topics

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Sodium fluoride is a chemical that has positively charged sodium ions bonded with negatively charged fluoride ions. It exists as a dissolvable white powder. If you remember the Periodic Table of Elements from your high school biology class, you'll understand right away why its chemical structure is abbreviated as "NaF."

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